INTRODUCTION

Nutrition is a chemical bond required by the body to perform its functions, which produce energy, build and maintain tissues, and regulate the processes of life. This proves that nutrition is a component that is needed by humans in their daily activities. Every human activity requires energy, both in small quantities and in large quantities. Therefore, PT. Novell Pharmaceutical Laboratories has the idea to design a beverage supplement that provides nutritional fulfillment for its consumers for 12 h after drinking it. This drink contains various vitamins and amino acids. This product is named Fitamino in the form of powder and has a taste of chocolate.

Fitamino is formulated in such a way as to meet the nutritional needs of consumers for 12 h. Therefore, testing of both vitamin and amino acids should be tested. This test is needed to ensure that the formulation has been designed in accordance with the products made in order to meet the nutritional needs of consumers for 12 h.

Amino acid

The structure of amino acids comprises an amino group, a carboxyl group, amino group, and an R group bound to an atom C known as a carbon α. Cluster R is a branch chain that distinguishes an amino acid with other amino acids. Amino acids in neutral conditions (pH isoelectric) are in the form of bipolar ions (zwitter ion). In bipolar amino acids, amino groups get additional protons and dissociated carboxyl groups. Amino acids are divided into two groups, namely essential amino acids and non-essential amino acids. Essential amino acids are amino acids that cannot be produced in the body so they must be added in the form of food, while non-essential amino acids are the amino acids that can be produced in the body.

Branched Chain Amino Acids (BCAA)

Branched Chain Amino Acid consists of leucine, isoleucine and valine is an amino acid that is metabolized in the muscle. In muscle, BCAA serves as an important energy substrate during exercise and stress periods, besides BCAA also acts as a precursor for the synthesis of amino acids and proteins.

BCAA is a contributor of 35-40% essential amino acid diet in the body protein and 14-18% of the total amino acid in muscle protein in which the human muscle mass is 40% of body weight.

Muscle proteins have a constant condition, where new proteins will continue to be produced while older proteins will be degraded. However, the new protein synthesis rate will be greater than the level of muscle protein damage. Consumption of supplements containing BCAA can maximize this condition in which BCAA can stimulate muscle protein synthesis.

Taurine

Taurine (2-aminoethanesulfonic acid) is a sulfur-containing amino acid that plays a role in several metabolic processes such as cardiac contraction and antioxidant activity. The presence of sulphonate components in taurine serves as opposed to other amino acid carboxyl groups so that the taurine has a value pKa

ABSTRACT

Fitamino product from PT. Novell Pharmaceutical is a product formulated in powder form. This product contains a variety of vitamins and amino acids that aim to meet the nutritional needs of consumers for 12 h. Amino acids contained in the product are Branched Chain Amino Acid (BCAA) consisting of isoleucine, leucine, valine and taurine. In this study, the measurement of amino acid levels was done using different instruments. The instruments used were HPLC (High Performance Liquid Chromatography) and UHPLC (Ultra High Performance Liquid Chromatography). The research was done on preparation stage of test sample and test of sample rate using HPLC and UHPLC. The difference between these two methods lied in the derivatisation process. In HPLC, derivatization process was done manually while in UHPLC derivatization, process was done automatically in UHPLC machine. The results of the test showed that the amino acid content tested by HPLC method was greater than the amino acid level tested by UHPLC method that were: taurine 102.2%; valine 102.6%; isoleucine 95.72%; and leucine 101.79% while in UHPLC method: taurine 79.164%; valine 67.763%; isoleucine 67.569%; and leucine 72.107%. The results on this UHPLC method were well below the 85-115% requirement. That was because there were different treatments during the test sample preparation process but the time required for the process to use UHPLC was faster than the HPLC method.

Keywords: Amino Acid, Branched Chain Amino Acid, Fitamino, HPLC, Novell Pharmaceutical, Taurine, UHPLC.

COMPARISON STUDY BETWEEN HPLC AND UHPLC METHODS FOR TESTING THE LEVEL OF AMINO ACID

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INTRODUCTION

Research Article

Comparison Study Between HPLC and UHPLC Methods for Testing the Level of Amino Acid

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ABSTRACT

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Keywords: Amino Acid, Branched Chain Amino Acid, Fitamino, HPLC, Novell Pharmaceutical, Taurine, UHPLC.
1.5 which makes taurine the most acidic amino acid. Taurine does not have the carboxyl groups necessary to form peptide bonds so it does not function as a builder of protein structures. Taurine is indispensable during growth and can be found in pure milk, eggs, meat, and fish, as well as food or beverage supplement products. Taurine is formed by the body in the liver followed by the oxidation reaction of the amino acid decarboxylation. Taurine serves to maintain the balance of membrane cells in active tissues such as brain tissue and heart. In addition, taurine also plays a role in several biological processes such as development of central nervous system (CNS) and retina, membrane stabilization, reproduction and immune system. Taurine is an amino acid that does not belong to proteins. Taurine is most prevalent in the brain, retina, muscle tissue, and organs throughout the body.

MATERIALS AND METHODS

Fitamino 1000 mg containing L-Valine, L-Isoleucine, L-Leucine, and Taurine; acetate buffer pH 7.0 (MERCK); NaOH 10 N (MERCK); Borat buffer pH 9.5 (MERCK); Acetic Acid 2% (MERCK); OPA solution (O-Phthalaldehyde) (MERCK); Reagent OPA (O-Phthalaldehyde) (MERCK); Motion Phase (Methanol: Acetate Buffer pH 7.0); Aspartate Acid (MERCK) solution (as an internal standard solution).

Preparation of Standard Solutions

Weighed working standard L-Valin for:
- Standard 80%: 20 mg
- Standard 100%: 25 mg
- Standard 120%: 30 mg

Weighed working standard L-Isoleucine for:
- Standard 80%: 20 mg
- Standard 100%: 25 mg
- Standard 120%: 30 mg

Weighed working standard L-Leucine for:
- Standard 80%: 40 mg
- Standard 100%: 50 mg
- Standard 120%: 60 mg

Weighed working standard L-Taurin for:
- 80% Standard: 80 mg
- Standard 100%: 100 mg
- Standard 120%: 120 mg

Each standard was put into a 100 mL measuring flask and dissolved with water and marked up with water until 100 mL.

Preparation of Test Solutions

Weighed 5 sachet Sustained Release Nutrition 12 h. The powder was carefully weighed equal to 1.9 g of sample and then put into a 50 mL measuring flask. After that, 25 mL of solvent was added, then sonicated and vortex. The solution was diluted with water until the marks were then centrifuged and filtered.

A. Examination Levels using HPLC

- Sample Preparation
  The sample was dissolved with hot water.

- Derivatization
  A total of 20 µL internal standard and 100 µL borate buffer solution pH 9.5 was piped. Then add 40µL of O-phthalaldehyde reagent, vortex for 3 min, wait 20 sec. Added 80µL Acetic Acid 2%, and 140 µL water, then shaken until homogeneous.

- HPLC condition
  Column: Purospher® RP-18e, 5µm, 250mm x 4.6mm
  Phase of motion: Phase of motion
  Detector: 338 nm
  Flow rate: 2.0 mL / min
  Input Volume: 10 mL
  Column Temperature: 40°C

B. Examination Levels using UHPLC

- Sample Preparation
  The sample was dissolved with plain water.

- Derivatization
  Into a vial was loaded water of 10 µL, 5 µL borate buffer, 1 µL sample solution. The vial was then inserted into the tool and mixed in the injector 3 times (6 µL each) and awaited for 15 sec. The OPA / MPA reagent solution was inserted into the vial and mixed back into the injector 6 times (each 7 µL and awaited for 15 seconds) The injector was washed Inserting the FMOC solution into the vial and mixed in the injector 6 times (each 8 µL) and then waited for 15 sec. The injector was washed and diluted inserted and mixed in injector 4 times (12 µL each) and awaited for 15 sec. Conducted a process of mocking and starting data calculation After the injector was emptied and washed with buffer loop.

RESULTS AND DISCUSSION

The results of analysis of amino acid levels using HPLC are shown in Table 1 to 8. The results of analysis of amino acid levels using UHPLC are shown in Table 9.

In this study, examination of amino acid levels was performed by using two different instruments. It aim was to analyze the effect of UHPLC usage on the effectiveness of work. The HPLC method was chosen because it has many advantages over the classical liquid chromatography method, such as being able to separate the molecules from a mixture, its execution is easy, the speed of analysis and high sensitivity, the resolution is good, can use various detectors, the columns that have been used can reused, and can avoid material damage.

The first step was to make standard solution and test solution. Standard solution and test solution were prepared by dissolving the substance with the solvent ie water. However, for test solution, testing was done using HPLC; the solvent used was hot water while in experiments using UHPLC, the solvent used was plain water. The second stage was process of derivatization. The process of derivatization was done so that amino acids could form derivatives that absorbed UV light, visible, or fluorescence. The compound used for derivatisation in this procedure was the O-phthalaldehyde reagent. This reagent could form an iso-indole derivative strongly fluorescing so it can be detected by a fluorescence detector. The derivatization process is of two kinds, namely post-column derivatization and precolumn derivatization. In this test, the method of derivatization of precolumn with separation mechanism was a partition with reverse phase chromatography (reverse phase chromatography). In this process the amino acids react selectively with OPA until a strongly hydrophobic fluorescent derivative was formed which enables the reverse phase chromatographic separation using nonpolar columns and a polar phase of motion. The mobile phase used in this test was methanol: buffer acetate. The derivatization process in the test using HPLC was done manually while the derivatization process in the test using UHPLC was done automatically in the UHPLC tool. After the derivatization process, both standard and test solutions were injected into the HPLC columns and tested.
Table 1. Result of Taurine Standard Level Examination

<table>
<thead>
<tr>
<th>Weight (mg)</th>
<th>Conc. (ppm)</th>
<th>Area IS</th>
<th>IS Ratio</th>
<th>Mean Area</th>
<th>Mean Area</th>
<th>SD</th>
<th>CV (%)</th>
</tr>
</thead>
<tbody>
<tr>
<td>100</td>
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<td>801380</td>
<td>373115</td>
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Table 2. Result of Taurine Sample Level Examination

<table>
<thead>
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<th>Sample</th>
<th>Area</th>
<th>IS</th>
<th>Ratio</th>
<th>Assay (%)</th>
<th>Mean Assay (%)</th>
<th>SD</th>
<th>CV (%)</th>
<th>Req.</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>808340</td>
<td>400056</td>
<td>2.021</td>
<td>94.33</td>
<td>102.2</td>
<td>11.14</td>
<td>10.00</td>
<td>85% - 115% CV ≤ 5%</td>
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<tr>
<td>2</td>
<td>895654</td>
<td>379821</td>
<td>2.358</td>
<td>110.08</td>
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Table 3. Valine Standard Level Examination Result

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<th>Weight (mg)</th>
<th>Conc. (ppm)</th>
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<th>IS Ratio</th>
<th>Mean Area</th>
<th>Mean Area</th>
<th>SD</th>
<th>CV (%)</th>
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<tbody>
<tr>
<td>25.02</td>
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<td>232662</td>
<td>373115</td>
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<td>0.01</td>
<td>0.88</td>
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<td>250309</td>
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<td>0.612</td>
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Table 4. Examination Result of Valid Samples

<table>
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<tr>
<th>Sample</th>
<th>Area</th>
<th>IS</th>
<th>Ratio</th>
<th>Assay (%)</th>
<th>Mean Assay (%)</th>
<th>SD</th>
<th>CV (%)</th>
<th>Req.</th>
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</thead>
<tbody>
<tr>
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<td>0.592</td>
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<td>2</td>
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<td>379821</td>
<td>0.582</td>
<td>109.89</td>
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</tr>
</tbody>
</table>

Table 5. Result of Standard Isoleucine Level Examination

<table>
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<tr>
<th>Weight (mg)</th>
<th>Conc. (ppm)</th>
<th>Area IS</th>
<th>IS Ratio</th>
<th>Mean Area</th>
<th>Mean Area</th>
<th>SD</th>
<th>CV (%)</th>
</tr>
</thead>
<tbody>
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<td>0.551</td>
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Table 6. Result of Isoleucine Sample Level Examination

<table>
<thead>
<tr>
<th>Sample</th>
<th>Area</th>
<th>IS</th>
<th>Ratio</th>
<th>Assay (%)</th>
<th>Mean Assay (%)</th>
<th>SD</th>
<th>CV (%)</th>
<th>Req.</th>
</tr>
</thead>
<tbody>
<tr>
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<td>0.520</td>
<td>93.20</td>
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<td>3.72</td>
<td>85% - 115% CV ≤ 5%</td>
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<tr>
<td>2</td>
<td>230675</td>
<td>421265</td>
<td>0.548</td>
<td>98.23</td>
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</tbody>
</table>

Table 7. Result of Standard Leucine Level Inspection

<table>
<thead>
<tr>
<th>Weight (mg)</th>
<th>Conc. (ppm)</th>
<th>Area IS</th>
<th>IS Ratio</th>
<th>Mean Area</th>
<th>Mean Area</th>
<th>SD</th>
<th>CV (%)</th>
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</thead>
<tbody>
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<td>50.00</td>
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<td>422589</td>
<td>373115</td>
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<td>1.133</td>
<td>0.01</td>
<td>0.69</td>
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<td>456524</td>
<td>409212</td>
<td>1.116</td>
<td></td>
<td></td>
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</tr>
</tbody>
</table>
Based on test results using HPLC, obtained results were: taurine content of 102.20%; valine content of 102.60%; isoleucine levels of 95.72%; and leucine level of 101.79%. These results have been eligible at 85-115% with duration of about 30 min. Meanwhile, the test results obtained using UHPLC were: taurine content of 79.64%; valine content of 67.76%; isoleucine content of 67.57%; and leucine level of 72.11%. The results were not yet eligible at 85-115% with a long working time of about 20 min. The test results using UHPLC were not yet qualified because of the small possibility for error due to automatic derivatization process in machine.

**REFERENCES**


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